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## INFLUENCE OF SILICON CARBIDES ON THE STRUCTURE AND PROPERTIES OF NICKEL-PHOSPHORUS COMPOSITE COATINGS

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**Abstract.** The authors studied the structure, properties, and corrosion resistance in different acids of the nickel-phosphorus coatings with the dispersed silicon carbides after crystallization annealing in different modes. Crystallization onset temperatures after heating at rates of 1, 5, and 20 °C/min and the percentage of crystalline phases formed under isothermal conditions (nickel phosphide  $\text{Ni}_3\text{P}$  and nickel) were determined. It was determined that a high microhardness of more than 1000 HV is achieved in the composite nickel-phosphorus coating with dispersed particles of the silicon carbides also during prolonged low-temperature annealing, accompanied by crystallization with the formation of already insignificant (10 %) amounts of  $\text{Ni}_3\text{P}$ . The revealed dispersed  $\text{Ni}_3\text{P}$  located both inside the grains and along the boundaries of the grains make the main contribution to the increase in microhardness. Yield strength and tensile strength of coatings increase during crystallization annealing by only 12 – 15 MPa, and elongation drops to zero, due to the formation of the brittle  $\text{Ni}_3\text{P}$  compounds. Annealing with a short-term soaking at crystallization temperatures leads to the fact that the silicon carbides exhibit a barrier effect. This reduces the intensity of the formation of crystalline  $\text{Ni}_3\text{P}$  and corrosion resistance, while a long-term soaking at lower crystallization temperatures forms about 70 %  $\text{Ni}_3\text{P}$ , contributing to consistently high hardness and improved corrosion resistance. Corrosion resistance of the composite Ni-P coatings with the silicon carbides, regardless of heat treatment modes, is maximum in acetic and orthophosphoric acids at the 70 % nickel phosphide and minimum in nitric acid and its mixtures with other acids.

**Keywords:** amorphous coatings, nickel-phosphorus, silicon carbides, crystallization annealing, nickel phosphides, nickel, microhardness, plasticity, corrosion resistance, acids

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## ВЛИЯНИЕ КАРБИДОВ КРЕМНИЯ НА СТРУКТУРУ И СВОЙСТВА КОМПОЗИТНОГО НИКЕЛЬ-ФОСФОРНОГО ПОКРЫТИЯ

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**Аннотация.** Статья содержит исследования структуры, свойств и коррозионной стойкости в различных кислотах никель-фосфорных покрытий с дисперсными карбидами кремния после кристаллизационного отжига по различным режимам. Установлены температуры начала кристаллизации после нагрева со скоростями 1, 5, 20 °C/мин и процентное содержание образующихся в изотермических условиях кристаллических фаз (фосфида никеля  $\text{Ni}_3\text{P}$  и никеля). Определено, что высокая микротвердость более 1000 HV достигается в композитном никель-фосфорном покрытии с дисперсными частицами карбидов кремния также при длительном низкотемпературном отжиге, сопровождающемся кристаллизацией с образованием уже незначительных (10 %) количеств фосфида никеля. Выявленные дисперсные фосфиды никеля, располагающиеся как в теле, так и по границам зерен, вносят основной вклад в приращение микротвердости. Предел текучести и предел прочности покрытий увеличиваются при кристаллизационном отжиге всего на 12 – 15 МПа, а относительное удлинение падает до нуля, что обусловлено образованием хрупких соединений фосфида никеля. Отжиг с непродолжительными выдержками при температурах кристаллизации приводит к тому, что карбиды кремния проявляют барьерный эффект, снижая интенсивность образования кристаллического фосфида никеля и коррозионную стойкость, тогда как продолжительные выдержки при более низких температурах кристаллизации формируют порядка 70 %  $\text{Ni}_3\text{P}$ , способствуя стабильно высокой твердости и улучшенным показателям коррозионной стойкости. Коррозионная стойкость композитных покрытий Ni-P + карбиды кремния вне зависимости от режимов термообработки максимальная в уксусной и ортофосфорной кислотах при 70 % фосфида никеля и минимальная в азотной кислоте и ее смесях с другими кислотами.

**Ключевые слова:** аморфные покрытия, никель-фосфор, карбиды кремния, кристаллизационный отжиг, фосфиды никеля, никель, микротвердость, пластичность, коррозионная стойкость, кислоты

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## INTRODUCTION

Chrome and nickel plating are the main types of the metal coatings. Currently, the nickel coatings occupy a leading position in the technological process of protecting components [1]. Thus, the composite nickel-phosphorus coatings have a significant wear resistance [2; 3], high corrosion resistance [4 – 7], good adhesion [8] and decorative properties [1]. A promising method of hardening and protecting components of a simple shape is the method of laser cladding of the nickel-based coatings, providing high tribological properties at high temperatures (about 1000 °C) [9; 10]. Methods of obtaining coatings by supersonic electric arc spraying are being studied and improved [11]. According to [12; 13], the most commonly used nickel-phosphorus coatings containing about 10 wt. % of phosphorus in their composition are amorphous after application. A subsequent heat treatment eventually transforms such coatings into a crystalline state which provides the necessary level of properties.

The composite nickel-phosphorus coatings usually have a layered structure, in the surface layer of which there are dispersed particles (silicon carbides, titanium, zirconium, diamond micro-powder [15 – 18]), contributing to the increase in service characteristics. Currently, in the manufacture of critical components used in transportation of oil and gas, the composite nickel-phosphorus coatings with the dispersed silicon carbides are being introduced, which increase the service life of products.

The purpose of this study was to determine the phase composition of the composite nickel-phosphorus coatings with the silicon carbide particles, which provides a high microhardness of more than 1000 HV in combination with a high corrosion resistance in different aggressive media.

## RESEARCH MATERIALS AND METHODOLOGY

We applied a double-layer 60 µm thick nickel-phosphorus coating (Ni-P: 30 µm, Ni-P + silicon carbides: 30 µm), or a single-layer 60 µm thick Ni-P coating to a prepared 300×100 mm, 4 mm thick ground surface made of steel grade 09G2S (EU analog: MnSi5) using the electroless technology with the hypophosphite ions [19; 20]. The single-layer coating composition (% wt.) was as follows: 89.32 – 90.15 Ni; 9.71 – 10.14 P; 0.10 – 0.22 Si; 0.15 – 0.43 Cu. In addition, we also applied the coatings to 3 mm thick ground sheets made of stainless steel grade

08Cr18Ni10Ti (EU analog: X6CrNiTi18-10) in an electroless nickel plating bath. The coatings were subsequently separated by bending for further analysis.

We studied the crystallization of the separated coatings under a continuous heating in a neutral argon atmosphere at rates of 1, 5 and 20 °C/min using a Netzsch STA 449 F1 Jupiter simultaneous thermal analyzer. Then we used the curves obtained by differential scanning calorimetry (DSC) to determine crystallization onset temperatures and to evaluate thermal effects. Under isothermal conditions, the samples of the nickel coatings separated from the substrate were heat-treated according to specified conditions in a LOIP LF-15/11-G1 lab muffle furnace in an oxidizing atmosphere.

We measured the Vickers microhardness at a 100 g load applied to a polished surface of the samples by indenting a diamond indenter on a DuraScan-50 microhardness tester with the ECOS Workflow software. The test procedure was compliant with GOST R ISO 6507-1-2007 (ISO 6507-1:2005). The microhardness measurement error was ±35 HV. We produced flat 20×250×0.06 mm samples of the coating separated from the substrate for tensile testing at a 5 mm/min rate on an Instron electromechanical tensile testing machine with a force of 250 kN. Tensile strength and yield strength measurement error was ±5 MPa, and elongation measurement error was 0.1 %.

The coating resistance to extremely aggressive media was evaluated using the gravimetric method. During the test, the coating was immersed in concentrated acids or their solutions for 24 h at room temperature. Before and after the test, the samples were washed in ethyl alcohol, dried, and weighed with a VLR-200 lab scale ( $0,25 \cdot 10^{-3}$  g error). The coating weight loss was estimated as a percentage.

The structure of the initial coatings and the coatings after different annealing modes was studied using an Olympus GX-51 inverted microscope. The surface of the prepared sections was etched for 10 s in a mixture of concentrated nitric and acetic acids using the liquid-drop method.

We used a Jeol JSM-7001F Schottky emission scanning electron microscope with an Oxford INCA X-max 80 SDD detector for electron microscopic studies of the structure. The instrument determines the chemical composition of individual structural components and draws distribution patterns of different elements in them.

For X-ray diffraction studies, we used a DRON-4-07 diffractometers (iron anode radiation) and Rigaku Ultima IV (copper anode radiation). We applied

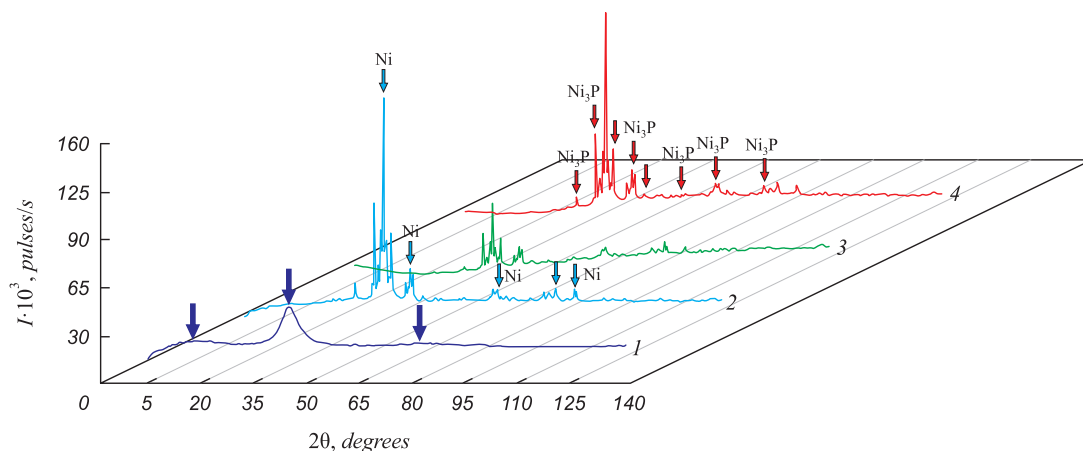


Fig. 1. X-ray diffractograms of the composite Ni-P coatings with the silicon carbides annealed at different temperatures and for different soaking time:

1 – without heat treatment; 2 – 450 °C (0.5 h); 3 – 420 °C (1 h); 4 – 390 °C (2 h)

Рис.1. Дифрактограммы композитных покрытий Ni-P + карбиды кремния, подвергнутых отжигу при различных температурах и времени выдержки:

1 – без т/о; 2 – 450 °C (0,5 ч); 3 – 420 °C (1 ч); 4 – 390 °C (2 ч)

the Rietveld method [21] for the qualitative and quantitative phase analysis after optimizing the interference peaks. The accuracy of the quantitative phase analysis was  $\pm 5\%$ . The sizes of the coherent scattering regions (CSRs) were determined by the Williamson–Hall and Halder–Wagner methods [22].

## RESEARCH RESULTS AND DISCUSSION

It was found that the coatings containing about 10 wt. % of phosphorus and about 1.0 % of the dispersed silicon carbide particles are in an amorphous state after application to steel substrates. The X-ray diffractograms of such coatings that were not subjected to heat treatment lack any interference peaks. There are only a few halos (marked with arrows) of different intensity in a wide range of  $2\theta$  reflection angles (Fig. 1, curve 1).

Microhardness of the initial Ni-P coatings is about 400 HV. As 1 % of the silicon carbide particles are added to the solution, it increases to 600 HV. This is also lower than the values of 1000 HV required by the specifications [23]. After application, the strength and plasticity properties of the coatings (refer to Table 1) are low (elongation varies from 0 to 1.5 %).

Heating the nickel-phosphorus coatings leads to crystallization and microhardness increase to the required values (over 1000 HV).

Crystallization onset temperature of the composite Ni-P coating with the silicon carbide particles is above 300 °C during a continuous heating. It largely depends on the heating rate, while a significant exothermic thermal effect changes insignificantly (Fig. 2).

As the samples are heated above the crystallization temperature, the X-ray diffractograms show diffraction peaks (see Fig. 1), indicating the formation of crystalline

phases in the coating. As our tests showed, after crystallization, in addition to the SiC and  $\text{Si}_3\text{C}_2$  carbides, the coating contains crystallized nickel, and precipitated  $\text{Ni}_3\text{P}$  compound. The coating structure is homogeneous and fine-grained. The grain size is 6 – 14  $\mu\text{m}$ , and the silicon carbide particle size is 0.5 – 1.5  $\mu\text{m}$  (Fig. 3, a).

The crystallization of the Ni-P coating with the silicon carbides develops noticeably under isothermal conditions at a temperature lower than during a continuous heating. For example, it was found that the coatings on a steel substrate contain about 10 % of  $\text{Ni}_3\text{P}$  after 24 h of soaking at 280 °C (refer to Table 2). If the temperature or annealing time is increased, the amount of the precipitated crystalline nickel phosphide reaches almost 70 % (refer to Table 2).

Note that for the coatings containing the dispersed silicon carbide particles, the proportion of these particles did not exceed 1 %. For this reason, we ignored them in the quantitative analysis of the phase composition in the calculation.

Table 1

### Mechanical properties of the Ni-P coatings with the silicon carbides separated from the substrate

Таблица 1. Механические свойства отделенных от подложки Ni-P покрытий с карбидами кремния

| Microhardness, HV     | Tensile strength, MPa | Yield strength, MPa | $\delta$ , % |
|-----------------------|-----------------------|---------------------|--------------|
| before heat treatment |                       |                     |              |
| 600                   | 177                   | 172                 | 0 – 1.5      |
| after heat treatment  |                       |                     |              |
| 1012 – 1080           | 189                   | 187                 | 0            |

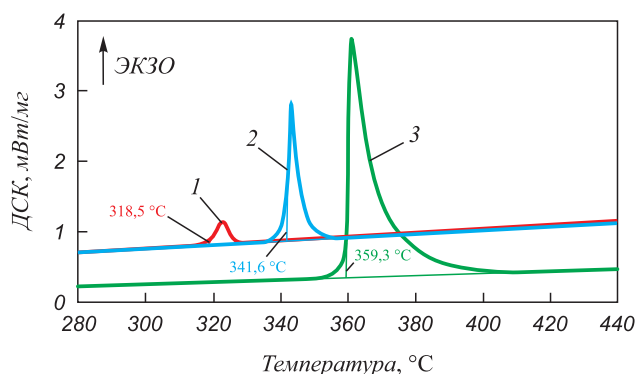


Fig. 2. DSC curves for heating in argon at the following rates:  
1 – 1 °C/min (89.1 J/g); 2 – 5 °C/min (87.2 J/g);  
3 – 20 °C/min (87.3 J/g)

Рис. 2. Кривые ДСК, полученные при нагреве в аргоне со скоростями:  
1 – 1 °C/мин (89,1 Дж/г); 2 – 5 °C/мин (87,2 Дж/г);  
3 – 20 °C/мин (87,3 Дж/г)

With the increase in the content of the nickel phosphide in the coating composition after 15 min of soaking at a temperature of 390 °C, microhardness increases from an initial 600 HV to an average of 976 HV, and to 1057 HV after 120 min of soaking, when 71 % Ni<sub>3</sub>P is formed (refer to Table 3).

When temperature reaches 420 °C, a similar high hardness of the samples is achieved in a shorter time. For all soaking ranges at the specified temperature, hardness is ensured with consistently obtained values of more than 1000 HV.

As the annealing temperature reaches 450 °C, microhardness is at its maximum after 30 min of soaking. Then it decreases due to phase coagulation and phosphorus burnout from the surface which gets a characteristic bluish hue. For a 24 h soaking period at 280 °C, the formed structure creates the same high hardness

Table 2

SCR size, amount of crystalline nickel ( $C_{Ni}$ ), and percentage of nickel phosphide ( $C_{Ni_3P}$ ) in the coating after different heat treatment modes

Таблица 2. Размер ОКР, количество кристаллического никеля ( $C_{Ni}$ ) и процентное содержание фосфида никеля ( $C_{Ni_3P}$ ) в покрытии после различных режимов т/о

| Annealing mode | SCR size, nm | $C_{Ni}$ , % | $C_{Ni_3P}$ , % |
|----------------|--------------|--------------|-----------------|
| 280 °C – 24 h  | 13.5         | 89           | 10              |
| 390 °C – 1.0 h | 20.9         | 75           | 24              |
| 390 °C – 2.0 h | 18.7         | 28           | 71              |
| 420 °C – 0.5 h | 16.0         | 78           | 21              |
| 420 °C – 1.0 h | 15.9         | 34           | 65              |
| 420 °C – 2.0 h | 25.5         | 32           | 67              |
| 450 °C – 0.5 h | 18.4         | 68           | 31              |

$HV_{avg} = 1016$  (1033; 1004; 1023; 1033; 985) HV.

The coatings with a high microhardness after heat treatment have low yield strength and tensile strength. These values are increased by only 12 – 15 MPa. Heat treatment makes the coatings so brittle that their plasticity drops to zero. Such changes in the coating properties during heat treatment are mainly due to the formation of the brittle nickel phosphide compounds with a high microhardness.

The phosphorus microvolume distribution chart helped us to identify the locations of nickel phosphides. Phosphorus in this coating is part of the Ni<sub>3</sub>P compound. Thus, the locations of phosphorus localization indicate that the nickel phosphides are located both inside the grains and along the grain boundaries. After annealing for 1 h at 420 °C, the nickel phosphide precipitates at all grain boundaries. The nickel phosphides inside the grains and along the grain boundaries are indicated by arrows in the electron microscope image (Fig. 3, b).

The fine structure of the coatings after crystallization annealing in different modes shows that the SCR sizes determined by the Williamson–Hall and Halder–Wagner methods are close. In crystalline nickel, they vary from 10 to 25 nm (refer to Table 2), while in the nickel phosphide they are slightly larger and range from 15 to 30 nm. As listed in Table 2, the lower the annealing temperature and the shorter the soaking time, the smaller the size of the SCR formed.

Corrosion resistance of the coatings to the effects of different aggressive media (acids and their solutions) is the most important quality metric along with a high hardness. It was found that the studied Ni-P coating with the silicon carbides has the highest corrosion resistance to acetic and orthophosphoric acids, regardless of heat treatment. After daily tests, the most aggressive medium for such coatings is nitric acid, its mixture with other acids, or even its solution diluted with distilled water. The coatings are completely dissolved in nitric acid and its solutions during daily tests (Fig. 4, curve 5). On the other

Table 3

Microhardness of the Ni-P coatings with the silicon carbides after annealing at different temperatures and for different soaking time ( $\tau$ )

Таблица 3. Микротвердость покрытий Ni-P + карбиды кремния после отжига при различных температурах и времени выдержки ( $\tau$ )

| $\tau$ , min | Microhardness, $HV_{avg}$ |        |        |
|--------------|---------------------------|--------|--------|
|              | 390 °C                    | 420 °C | 450 °C |
| 15           | 976                       | 1047   | 1002   |
| 30           | 989                       | 1049   | 1061   |
| 60           | 1047                      | 1062   | 1021   |
| 120          | 1057                      | 1039   | 1018   |

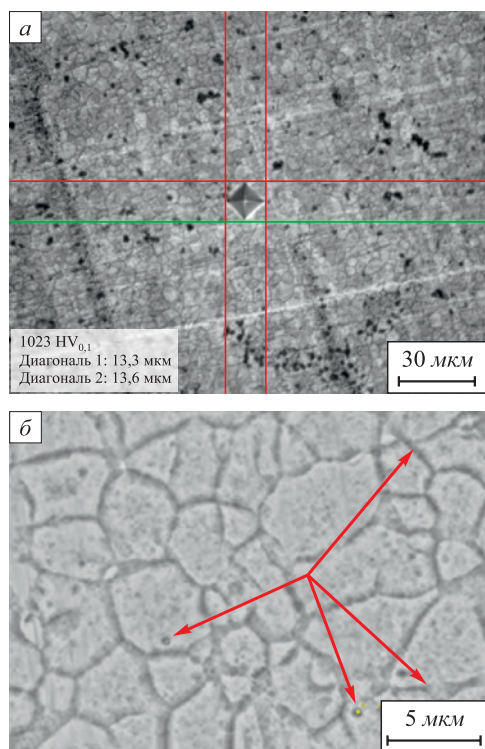


Fig. 3. Microstructure of the Ni-P coating with the silicon carbides after annealing at 420 °C for 1 h:  
a –  $\times 400$ ; б – SEM,  $\times 5000$

Рис. 3. Микроструктура покрытия Ni-P + карбиды кремния после отжига при 420 °C в течение 1 ч:  
a –  $\times 400$ ; б – РЭМ,  $\times 5000$

hand, the max daily weight loss during soaking in sulfuric acid is less and amounts to 5.3 %. The weight loss in hydrochloric acid is 11.2 %, which is also less than in nitric acid.

We also found a relationship between corrosion resistance of the Ni-P coatings with the silicon carbides and their quantitative phase composition. Corrosion resistance is at its maximum after 2 h of soaking at 390 °C (Fig. 4) or 1 h of soaking at 420 °C when about 70 %  $\text{Ni}_3\text{P}$  is formed. Its stability is greater than that of pure nickel or its other compounds with phosphorus. The presence of the dispersed silicon carbides decreases the weight loss of the samples compared to the pure Ni-P coating [24]. The introduction of the silicon carbide as a dispersed phase, which creates a barrier effect for the formation of the nickel phosphides, makes it possible to achieve the set goals with the use of a longer treatment at lower temperatures. This also improves main service characteristics of the coating such as a high microhardness and corrosion resistance.

## CONCLUSIONS

The required 1000 HV of microhardness according to the specifications is achieved in the composite nickel-phosphorus coating with the dispersed silicon carbide par-

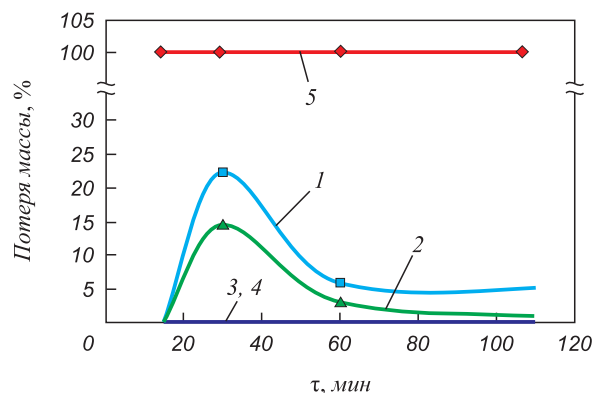


Fig. 4. Weight loss (%) vs. heat treatment period at 390 °C for the Ni-P coatings with the silicon carbides after a 24 h exposure to concentrated hydrochloric (1), sulfuric (2), acetic (3), orthophosphoric (4) and nitric (5) acids

Рис. 4. Потеря массы в зависимости от времени т/о при 390 °C Ni-P покрытий с карбидами кремния после их суточной выдержки в концентрированных соляной (1), серной (2), уксусной (3), ортофосфорной (4) и азотной (5) кислотах

ticles during a long-term low-temperature annealing with crystallization and the formation of insignificant (10 %) amounts of nickel phosphide.

The revealed dispersed nickel phosphides formed during crystallization and located both inside the grains and along the grain boundaries make a main contribution to the microhardness increase.

A max corrosion resistance of the coatings to different acids combined with a high microhardness is achieved at a high (70 %) nickel phosphide content. Its amount increases with the increase in annealing temperature or time.

With a significant increase in microhardness of the coatings from the initial 600 to the required 1000 HV after crystallization annealing, yield strength and tensile strength increase by only 12 – 15 MPa, and elongation drops to zero due to the formation of brittle nickel phosphide compounds.

Heat treatment of the Ni-P coatings with the silicon carbides forms a homogeneous, fine-grained structure with a grain size of 6 – 14 μm and the SCR size of 10 – 25 nm in nickel, and 15 – 30 nm in the nickel phosphide.

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